

Original Research Article

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Effluent Quality of a Carbonated Soft Drink (CSD) Company in Owerri, Imo State, Nigeria and the Receiving Water Body

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ABSTRACT

The effluent quality of a carbonated soft drink (CSD) company in Owerri and the receiving water body (Nworie River) were determined using standard microbiological and physicochemical methods. Microbial loads showed that total heterotrophic count (THC) ranged from $1.0 \times 10^2 \pm 0.14$ to $3.7 \times 10^5 \pm 0.53$ CFU/ml, total coliform count (TCC) ranged from $0.2 \times 10^1 \pm 0.19$ to $2.9 \times 10^5 \pm 0.50$ CFU/ml and yeast and mould count (Y&MC) ranged from $1.1 \times 10^2 \pm 0.1$ to $2.1 \times 10^4 \pm 0.93$ CFU/ml ($p < 0.05$). Six (6) bacterial isolates were identified to include *Pseudomonas* species, *Bacillus* species, *Staphylococcus* species, *Escherichia coli*, *Flavobacterium* species, and *Azotobacter* species, while six (6) fungal isolates were identified to include *Aspergillus* species, *Penicillium* species, *Acremonium* species, *Rhodotorula* species, *Saccharomyces cerevisiae*, and *Candida* species. Physicochemical results obtained in this study showed a pattern ($p < 0.05$) with higher values recorded in the influent and least values in the effluent samples, while in the receiving water body, higher values were recorded in the discharge point, followed by downstream and least values in the upstream samples. However, it was discovered that discharges into the same tunnel from neighborhood companies around the industrial site increased the pollution level of the discharged effluent into the river, leaving potential effects on the downstream. Having considered the need for a sustainable water economy with looming water scarcity globally, and associated health hazards with untreated wastewater, there is need for proper monitoring by concerned agencies in ensuring wastewater from industries are treated to meet minimum permissible quality standards to avoid midstream contamination.

Keywords

Effluent, Receiving water body, Microbiological, Physicochemical, Pollution level

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Introduction

Nigeria is the most populous country in Africa with a population of over 160 million people; the country is endowed with generous resources of water bodies. This water provides resources for fishery, transportation, irrigation, recreation and domestic use (Ekiye and Luo, 2010). There are different regulations put in place to protect the marine environment and other water bodies in Nigeria. However, they have not been effective in controlling the indiscriminate dumping of effluent into open water bodies.

Inability to effectively and efficiently manage huge amount of wastes generated by various anthropogenic activities particularly in developing countries has created serious problems in our environment. Industrial effluent contamination of natural water bodies has emerged as a major challenge in developing and densely populated countries like Nigeria (Sangadoyin, 1995). River systems are the primary means for disposal of waste, especially the effluents, from industries that are near them. These effluent from industries have a great deal of influence on the pollution of the water body, these effluent can alter the physical, chemical and biological nature of the receiving water body (Sangadoyin, 1991). Increased industrial activities have led to pollution stress on surface waters both from industrial, agricultural and domestic sources (Ajayi and Osibanjo, 1981). The manner at which industrial effluents are being disposed into the environment (water bodies) like rivers, streams, etc. are becoming worrisome.

Over the years, in many African countries a considerable population growth has taken place, accompanied by a steep increase in urbanization and industrialization. This has led to tremendous increase in discharge of wide diversity of pollutants to receiving water

bodies (Saad *et al.*, 1985). The discharge of industrial effluents into receiving water bodies in Nigeria invariably result in the presence of high concentrations of pollutant in the water and sediment. The pollutants have been present in concentrations, which may be toxic to different organisms. The effluents also have considerable negative effects on the water quality of the receiving water bodies and as such, they are rendered unsafe for human use (Nwokorie and Ike, 2016; Kanu and Achi, 2011). The consequences of these negative effects are of great health concern (Osibanjo *et al.*, 2011). One of the most important factors of water pollution is the microbial contamination, especially with pathogenic microorganisms. Enteric pathogens are typically responsible for several waterborne sicknesses (Sabae, 2004). However, these water bodies as in the case of developing countries like Nigeria serve as major source of water for domestic uses. The inhabitants depend on this available source for their daily drinking water and other activities like washing of clothes, bathing, etc. (Sangadoyin, 1995). Ideally, effluents from industries are to be properly treated before being discharged into the environment. In Nigeria, the case is different as there are many sharp practices by industrialists in managing effluent. Therefore, this study is targeted to assess the effluent quality of a carbonated soft drink company in Owerri and the receiving water body (Nworie River).

Materials and Methods

Study area

The study area is Owerri-West Local Government Area, in Owerri, Imo State, Nigeria. The geographical coordinates are 5.1215⁰N, and 7.3732⁰E. The area is of tropical climatic conditions with rain forest features. The soil type is silt-clay and the weather is typical of rain forest, with average

annual temperature ranging between 25 and 35°C as lowest and highest values, respectively.

Sample sources and collection

Samples were collected from five (5) different sampling points which include Wastewater Treatment Plant (WWTP) and receiving water body (Nworie River). Two out of the five sampling points (influent and effluent) were from the wastewater treatment plant (also known as Effluent Treatment Plant - ETP), while the other three samples (upstream, discharge point and downstream) were from specified points in the receiving water body. Samples were collected in duplicate from the five sampling points. These samples were packaged aseptically for microbiological and physicochemical analysis. Universal sample bottles (sterile) of 100ml and 1000ml capacities were used for microbiological and physicochemical sampling respectively. The samples were transported in an icebox with sufficient ice blocks to maintain temperature of about 4 - 6°C. The samples were then stored at 4°C at the refrigerator in the laboratory until use. All wastewater samples for microbiological analysis were analyzed within twenty four (24) hours after collection. The water samples for physicochemical analysis were analyzed within one (1) week of collection.

Microbiological studies

Tenfold serial dilutions of samples were done using sterile peptone water as the diluent. Spread plate and streaking techniques were used to enumerate and isolate bacteria and fungi in the samples (Cappucino and Sherman, 2010). Sample dilutions were prepared by adding 1.0ml of wastewater stock sample to 9.0ml of the sterile peptone water with vigorous agitation in ensuring adequate disengagement of microorganisms to obtain

10⁻¹ dilution. Serial dilutions of the homogenates were continued aseptically and made stepwisely till the sixth (6th) tube, to obtain dilutions of 10⁻³ to 10⁻⁶ dilutions and appropriate dilutions were plated in replicates using plate count agar for mean aerobic bacteria enumeration and isolation, tergitol agar for coliform enumeration and isolation, fortified sabouraud dextrose agar (SDA) for fungal enumeration and isolation. Pure bacterial isolates were identified using cultural, morphological and biochemical characterization. Identification of the bacteria to genera level was based on the schemes of Boone *et al.*, (2005). The purified fungal isolates were identified on the basis of macroscopic and microscopic characteristics by slide culture technique, and lactophenol staining. The schemes of Barnet and Hunter (2000) and Watanabe (2010) were used for the identification. The plates were incubated at 35 ± 2°C for 72 hours and 24 hours for total bacterial and coliform counts respectively and 25 ± 2°C for 120 hours for fungal counts.

Physicochemical studies

All the physicochemical parameters namely: appearance (apparent colour), odour, colour (true colour units), temperature °C (*in situ*), differential temperature (discharge & stream), pH, turbidity (NTU), conductivity (µs/cm), total hardness (CaCO₃) mg/l, dissolved oxygen (DO) mg/l, total solid mg/l, total suspended solids (TSS) mg/l, total dissolved solids (TDS) mg/l, biochemical oxygen demand (BOD₅) mg/l, chemical oxygen demand (COD) mg/l, acidity (CaCO₃, mg/l), alkalinity (CaCO₃, mg/l), chloride mg/l, nitrate (NO₃⁻) mg/l, sulphate (as SO₄²⁻) mg/l, total phosphorus mg/l, free chlorine mg/l, chromium (hexavalent) mg/l, lead mg/l, nitrogen total mg/l, nickel mg/l, aluminum mg/l, manganese mg/l, cadmium (mg/l), cobalt mg/l, copper mg/l, total chromium mg/l, zinc mg/l, ammonia as nitrogen mg/l, Iron mg/l, calcium

mg/l, magnesium mg/l, oil and grease mg/l, surfactants reacting to methylene blue were determined using methods described in CBM (2001) and WTM (1999).

Data analysis

Analysis of variance (ANOVA) was employed in this work and used to analyze all data obtained from the determinations. Descriptive statistics in form of mean and standard deviation and Duncan post hoc were also used to assess the data. The analyses were done using (Statistical Product and Service Solutions) SPSS 16.

Results and Discussion

The microbial diversity associated with a carbonated soft drink (CSD) wastewater in Owerri (influent and effluent) and the receiving water body – Nworie river (upstream, discharge point, and downstream) are shown in Table 1. From the results obtained, all the microbial counts enumerated were higher in the influent (THC- $2.0 \times 10^5 \pm 0.87^a$, TCC- $1.3 \times 10^5 \pm 0.35^a$, Y&MC- $1.5 \times 10^5 \pm 0.35^a$) and lower in the effluent (THC- $1.0 \times 10^2 \pm 0.14^e$, TCC- $0.2 \times 10^1 \pm 0.19^e$, Y&MC- $1.1 \times 10^2 \pm 0.10^d$). There were significant differences ($p < 0.05$) in microbial counts among the various groups: CSD wastewater (influent and effluent) and the receiving water body (upstream, discharge point and downstream).

In the receiving water body – the river, there were significant differences ($p < 0.05$) among the samples (upstream - THC- $3.7 \times 10^5 \pm 0.53^{ab}$, TCC- $2.9 \times 10^5 \pm 0.50^{ab}$, Y&MC- $2.1 \times 10^4 \pm 0.93^b$, discharge point - THC- $4.8 \times 10^3 \pm 0.31^d$, TCC- $2.1 \times 10^2 \pm 0.18^d$, Y&MC- $1.4 \times 10^2 \pm 1.02^d$, and downstream - THC- $2.1 \times 10^4 \pm 0.22^c$, TCC- $1.9 \times 10^4 \pm 0.47^c$, Y&MC- $2.2 \times 10^3 \pm 0.66^c$). However, no significant difference ($p > 0.05$) between effluent sample ($1.1 \times 10^2 \pm 0.10^d$) and that of discharge point

($1.4 \times 10^2 \pm 1.02^d$) under yeast and mold counts.

The results of the physicochemical parameters namely: appearance (apparent colour), odour, colour (true colour units), temperature $^{\circ}\text{C}$ (*in situ*), differential temperature (discharge and stream), pH, turbidity (NTU), conductivity ($\mu\text{s}/\text{cm}$), total hardness (CaCO_3), mg/l, dissolved oxygen (DO) mg/l, total solid mg/l, total suspended solids (TSS) mg/l, total dissolved solids (TDS) mg/l, biochemical oxygen demand (BOD_5) mg/l, chemical oxygen demand (COD) mg/l, acidity (CaCO_3 , mg/l), alkalinity (CaCO_3 , mg/l), chloride mg/l, nitrate (NO_3^-) mg/l, sulphate (as SO_4^{2-}) mg/l, total phosphorus mg/l, free chlorine mg/l, chromium (hexavalent) mg/l, lead mg/l, nitrogen total mg/l, nickel mg/l, aluminum mg/l, manganese mg/l, cadmium (mg/l), cobalt mg/l, copper mg/l, total chromium mg/l, zinc mg/l, ammonia as nitrogen mg/l, Iron mg/l, calcium mg/l, magnesium mg/l, oil and grease mg/l, surfactants reacting to methylene blue of the various sample groups were clearly tabulated in Table 2. Some of the results obtained from influent samples were above permissible limits (odour – unpleasant, pH - 9.73 ± 1.02^a , turbidity - 10.00 ± 0.02^a , alkalinity - 218.65 ± 1.02^a , biochemical oxygen demand - 67.32 ± 1.07^a , chemical oxygen demand - 110.10 ± 0.22^a , total dissolved solids - 750.09 ± 0.02^a , total suspended solids - 32.15 ± 0.09^a and nitrate - 16.02 ± 0.02^a except for heavy metals. All other sample groups (effluent, upstream, discharge point and downstream) had their entire results within permissible limits ($p < 0.05$). The influent from the carbonated soft drink (CSD) wastewater in Owerri was polluted to the extent that most microbiological and physicochemical parameters were significantly high ($p < 0.05$) and could be attributed to the heavy discharges from the various operational units within the factory.

Table.1 Bioloads of a carbonated soft drink (CSD) wastewater (influent and effluent) and the receiving water body – Nworie river (upstream, discharge point, and downstream)

MICROBES	COUNT (CFU/ml)				
	INFLUENT	EFFLUENT	UPSTREAM (US)	DISCHARGE POINT (DP)	DOWNSTREAM (DS)
THC	$2.0 \times 10^5 \pm 0.87^a$	$1.0 \times 10^2 \pm 0.14^e$	$3.7 \times 10^5 \pm 0.53^{ab}$	$4.8 \times 10^3 \pm 0.31^d$	$2.1 \times 10^4 \pm 0.22^c$
TCC	$1.3 \times 10^5 \pm 0.35^a$	$0.2 \times 10^1 \pm 0.19^e$	$2.9 \times 10^5 \pm 0.50^{ab}$	$2.1 \times 10^2 \pm 0.18^d$	$1.9 \times 10^4 \pm 0.47^c$
Y&MC	$1.5 \times 10^5 \pm 0.35^a$	$1.1 \times 10^2 \pm 0.10^d$	$2.1 \times 10^4 \pm 0.93^b$	$1.4 \times 10^2 \pm 1.02^d$	$2.2 \times 10^3 \pm 0.66^c$

Values are given as mean \pm SD (standard deviation). Within rows, values followed by the same alphabets are not significantly different but those followed by different alphabets are significantly different. Legend: THC= Total heterotrophic count, CC= Coliform count, Y&MC= Yeast and mold count. NESREA, (2009).

Table.2 Physicochemical results of a carbonated soft drink (CSD) wastewater (influent and effluent) and the receiving water body – Nworie river (upstream, discharge point, and downstream)

S/N	PARAMETERS	RESULTS OF ANALYSIS				
		Influent	Effluent	Upstream (US)	Discharge point (DP)	Downstream (DS)
	PHYSICO-CHEMICAL TESTS:					
1.	Appearance (apparent colour)	Slightly cloudy	Clear	Slightly clear	Slightly clear	Slightly clear
2.	Odour	Unpleasant	Odourless	Odourless	Odourless	Odourless
3.	Colour (True colour units)	15.00 ± 0.89^a	3.00 ± 0.16^b	7.00 ± 0.43^c	6.00 ± 0.08^c	7.00 ± 0.21^c
4.	Temperature $^{\circ}C$ (<i>in situ</i>)	28.70 ± 0.07^a	25.40 ± 0.22^b	26.90 ± 0.04^{bc}	27.20 ± 0.60^c	27.04 ± 0.17^c
5.	Differential Temperature (Discharge & stream)		1.64 ± 0.14^a			
6.	pH	9.73 ± 1.02^a	7.90 ± 0.09^b	8.28 ± 1.02^c	8.56 ± 0.31^c	8.43 ± 0.05^c
7.	Turbidity NTU	10.00 ± 0.02^a	4.00 ± 0.21^b	5.04 ± 0.99^c	5.27 ± 0.65^c	5.16 ± 0.72^c
8.	Conductivity (μ s/cm)	1277 ± 0.66^a	832 ± 0.50^b	506.90 ± 0.14^d	919.02 ± 0.66^c	596.40 ± 0.70^d
9.	Total Hardness (CaCO ₃ , mg/l)	62.50 ± 0.89^a	50.0 ± 0.91^b	47.50 ± 0.16^c	60.03 ± 0.11^d	49.62 ± 0.53^b
10.	Dissolved Oxygen (DO) mg/l	2.89 ± 0.16^{de}	3.75 ± 0.11^{bc}	5.07 ± 0.09^a	3.15 ± 0.74^d	4.08 ± 0.22^b
11.	Total Solid mg/l	812.03 ± 0.45^a	426.07 ± 0.71^c	304.90 ± 0.05^{de}	486.32 ± 0.65^b	331.55 ± 0.81^d
12.	Total Suspended Solids (TSS) mg/l	32.15 ± 0.09^a	10.53 ± 0.03^{bc}	6.34 ± 0.20^d	13.41 ± 0.19^b	6.49 ± 0.04^d
13.	Total Dissolved Solids (TDS) mg/l	750.09 ± 0.02^a	405.04 ± 0.07^c	214.01 ± 0.02^e	457.06 ± 0.12^b	289.22 ± 0.17^d
14.	Biochemical Oxygen Demand (BOD ₅) mg/l	67.32 ± 1.07^a	33.7 ± 0.66^c	25.12 ± 0.30^d	40.3 ± 0.07^b	31.10 ± 0.42^c
15.	Chemical Oxygen Demand (COD) mg/l	110.10 ± 0.22^a	56.50 ± 0.52^b	37.33 ± 0.23^{cd}	58.17 ± 0.18^b	41.25 ± 0.10^c
16.	Acidity (CaCO ₃ , mg/l)	35.00 ± 0.64^a	17.50 ± 0.88^c	15.58 ± 0.54^{cd}	19.01 ± 0.65^b	17.50 ± 0.43^c
17.	Alkalinity (CaCO ₃ , mg/l)	218.65 ± 1.02^a	120.07 ± 0.99^b	114.60 ± 0.76^{bc}	122.04 ± 0.12^b	119.55 ± 0.66^b
18.	Chloride mg/l	54.12 ± 1.01^a	50.02 ± 1.20^a	26.59 ± 0.19^{bc}	54.90 ± 0.14^a	38.67 ± 1.01^b
19.	Nitrate (NO ₃ ⁻) mg/l	16.02 ± 0.02^a	4.83 ± 0.33^c	3.20 ± 0.09^{cd}	5.62 ± 0.22^b	4.45 ± 0.03^c
20.	Sulphate (as SO ₄ ²⁻) mg/l	79.91 ± 0.55^a	71.00 ± 0.87^c	26.00 ± 0.55^e	75.05 ± 0.66^b	40.50 ± 0.08^d
21.	Total Phosphorus mg/l	0.80 ± 0.06^a	0.53 ± 0.12^b	0.21 ± 0.18^{cd}	0.59 ± 0.05^b	0.33 ± 0.04^c
22.	Free Chlorine mg/l	0.25 ± 0.02^a	ND	ND	ND	ND
23.	Chromium (hexavalent) mg/l	0.003 ± 0.55^a	0.001 ± 0.19^a	BDL	0.001 ± 0.13^a	BDL
24.	Lead mg/l	0.002 ± 0.04^a	0.001 ± 0.05^a	ND	0.001 ± 0.02^a	BDL
25.	Nitrogen Total mg/l	10.63 ± 0.12^a	6.18 ± 0.47^b	1.22 ± 0.02^{de}	4.18 ± 0.47^c	2.60 ± 0.64^d
26.	Nickel mg/l	0.003 ± 0.01^a	0.002 ± 0.11^a	BDL	0.001 ± 0.31^a	BDL
27.	Aluminum mg/l	0.04 ± 0.09^a	0.02 ± 0.15^{ab}	BDL	0.01 ± 0.14^c	BDL
28.	Manganese mg/l	0.01 ± 0.02^a	0.01 ± 0.10^a	BDL	0.005 ± 0.13^b	BDL
29.	Cadmium (mg/l)	0.24 ± 0.13^a	0.11 ± 0.05^b	BDL	0.005 ± 0.82^c	BDL
30.	Cobalt mg/l	0.003 ± 0.22^a	0.001 ± 0.41^b	BDL	0.001 ± 0.19^b	BDL
31.	Copper mg/l	0.13 ± 0.76^a	0.11 ± 0.42^a	BDL	0.005 ± 0.05^b	BDL
32.	Total Chromium mg/l	0.37 ± 0.44^a	0.21 ± 0.76^b	BDL	0.01 ± 0.54^c	BDL
33.	Zinc mg/l	0.12 ± 0.03^a	0.10 ± 0.01^a	BDL	0.005 ± 0.10^c	BDL
34.	Ammonia as Nitrogen mg/l	0.24 ± 0.55^a	0.18 ± 0.34^b	BDL	0.01 ± 0.22^c	BDL
35.	Iron mg/l	0.02 ± 0.21^a	0.01 ± 0.19^a	BDL	0.01 ± 0.10^a	BDL
36.	Calcium mg/l	32.21 ± 0.77^a	20.06 ± 0.55^c	9.04 ± 0.31^d	25.12 ± 0.28^b	11.45 ± 0.11^d
37.	Magnesium mg/l	68.24 ± 0.22^a	61.92 ± 0.87^b	16.32 ± 0.20^{de}	58.80 ± 0.15^{bc}	21.03 ± 0.65^d
38.	Oil and Grease mg/l	4.30 ± 1.06^a	ND	ND	ND	ND
39.	Surfactants reacting to methylene blue	2.24 ± 0.02^b	2.62 ± 1.18^a	0.52 ± 0.15^e	2.12 ± 0.78^{bc}	1.01 ± 0.21^d

Values are given as mean \pm SD (standard deviation). Within rows, values followed by the same alphabets are not significantly different but those followed by different alphabets are significantly different. Legend: ND = Not Detected, BDL = Below Detection Limit, NESREA, (2009).

The high values recorded in total dissolved solids and suspended solids were as a result of spills and discharges of chemicals during operations. Some of these pollutants served as nutrient to most inherent degrading microbes. The sugar spills from the syrup room were major source of carbon to inhabiting microorganisms. This contributed to the high bioloads recorded in the results. According to Osibanjo *et al.*, (2011); Gray (2009); and, Ekhaise and Ayansi (2005) the high microbial loads reflected the level of pollution in the environment which indicates the amount of organic matter and toxic substances present. The results obtained in this study correlated with the assertion of Osibanjo *et al.*, (2011); Gray (2009); and, Ekhaise and Ayansi (2005).

Microorganisms obtained in this study include bacteria: *Pseudomonas* species, *Bacillus* species, *Staphylococcus* species, *Escherichia coli*, *Flavobacterium* species, *Azotobacter* species, and fungi: *Aspergillus* species, *Penicillium* species, *Acremonium* species, *Rhodotorula* species, *Saccharomyces cerevisiae*, *Candida species*. The microbes obtained in this study were in agreement with the research reports of Nwokorie and Ike (2016) and Ibekwe *et al.*, (2004).

The presence of coliforms in the carbonated soft drink (CSD) wastewater samples (influent and effluent) could be traced to poor sanitary practices among workers in the plant and is an indication of fecal contamination, while that of the receiving water body (upstream and downstream) could be due to either discharges into the river and/or from activities of users of the stream. Activities of the river/ stream users include swimming, washing of clothes, bathing, and the crude practice of urinating and defecating into water body. During rainy season, the run offs from stormwater equally contributes to the contamination evidenced with receiving water body. The presence of *Escherichia coli* in

wastewater discharges is an indication of poor sanitary conditions in the practice of personal hygiene in food safety environment and could lead to faecal contamination of products. *Bacillus*, *Pseudomonas*, *Flavobacterium* and *Azotobacter* species are known to be environmental contaminants and can be found in the air, water and soil, while *Staphylococcus aureus* is known to inhabit the human skin as an opportunistic microorganism. *Bacillus* species are known as spore formers and can withstand harsh weather conditions. The presence of *Aspergillus*, *Penicillium*, *Acremonium* and *Rhodotorula* species could be attributed to the surrounding environment (Ike *et al.*, 2015). *Saccharomyces cerevisiae* was associated with sugar and sugar fermentations. The activities of making syrup resulting in sugar spills encouraged the replication of *Saccharomyces cerevisiae*. Also, beverage product spills and bursts during filling at the filler could contribute to sugar concentrations in the wastewater. Sugar is a major source of carbon, and as a nutritional element encourages growth of various groups of microorganisms including the degrading aerobes.

The pH of the influent was slightly alkaline (9.73 ± 1.02) and above set regulatory standard (NESREA) due to discharges of caustic soda during bottle washing and cleaning of process lines. It was discovered that pH values sometimes could shoot up to 10.26 during the monthly dumping of caustic from the washing machine. This adversely affects the microbial community and could be remedied by addition of hydrochloric acid at the equalization tank to neutralize the alkaline condition. When optimal pH is achieved, it encourages microbial activities especially in replication and degradation of organic matter to facilitate settleability of flocks as sludge to produce clear effluent. This was observed in the physicochemical results obtained in

sample of discharged effluent which invariably has met permissible limits as recommended by NESREA, and recorded in Table 2.

However, due to negative activities around the neighborhoods by some companies in the same industrial site that uses the same underground sewer tunnel in discharging their wastewater, it was observed that such discharges had significant pollution impact on the regulated discharges from the carbonated soft drink (CSD) company as evidenced in the discharge point (DP) sample results. These polluted discharges have seriously marred the concerted efforts of the CSD company in context, in maintaining wholesome effluent discharge requirements as recommended by NESREA. The after effect of these pollutions could range from pathogenic microbes (Okoh, 2007), eutrophication, depletion of oxygen (Kanu and Achi, 2011) to fouling of the water body (Osibanjo *et al.*, 2011). These could create uncondusive environment with unnecessary stress that affects aquatic lives and exposes users to different health hazards/ water borne diseases as reported by Nwokorie and Ike (2016); Kanu and Achi (2011); Osibanjo *et al.*, (2011); and Ekhaise and Ayansi (2005).

Having considered the need for a sustainable water economy with looming water scarcity globally, and associated health hazards with untreated wastewater, there is need for proper monitoring by concerned agencies in ensuring wastewater from industries are treated to meet minimum quality requirements for every intended use before they are discharged.

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